# organic papers

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#### **Key indicators**

Single-crystal X-ray study T = 193 K Mean  $\sigma$ (C–C) = 0.002 Å R factor = 0.044 wR factor = 0.120 Data-to-parameter ratio = 13.3

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

## 3-(1*H*-Indol-3-yl)-4-(3,4,5-trimethoxyphenyl)-2,5-dihydro-1*H*-pyrrole-2,5-dione

The crystal structure of the title compound,  $C_{21}H_{18}N_2O_5$ , was determined in order to study the electrocyclic reactivity of 3,4-diaryl-1*H*-pyrrole-2,5-dione derivatives. Intermolecular hydrogen bonds form sheets.

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## Comment

The title compound, (I), containing the maleimide (1*H*-pyrrole-2,5-dione) moiety as the core structure, was prepared in an attempt to synthesize analogues of combretastatin A-4, a well known antithrombotic compound (Woods *et al.*, 1995). Among the synthetic methods used for preparing asymmetrical 3,4-diarylmaleimides, the method of Faul *et al.* (1999) has been used. Purification was achieved by column chromatography, yielding (I) as yellow crystals. Crystals of (I) suitable for X-ray analysis precipitated at 278 K from an ethanol solution by slow evaporation; however, at room temperature in an ethyl acetate solution, compound (I) was found to be chemically unstable and generated the corresponding carbazole, (II), which is described in a subsequent report (Peifer *et al.*, 2005). The reaction scheme below shows the disrotatory cyclization of (I) and subsequent oxidation to yield (II).





A comparable mechanism for the reactivity of the class of 3,4-diarylmaleimides has been reported by Sanchez-Martinez *et al.* (2003) and Harris *et al.* (1993). The constitutions of maleimide (I) and carbazole (II) were investigated by <sup>1</sup>H and <sup>13</sup>C NMR spectroscopic analyses and were confirmed by X-ray analysis, showing the compounds to have the desired structures (Fig. 1). We now report the X-ray crystal structure analysis which definitively proves the structure of (I) and confirms the assumption of oxidative cyclization reactivity of

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*ORTEPII* view (Johnson, 1976) of a molecule of (I). Displacement ellipsoids are drawn at the 50% probability level. H atoms are depicted as circles of arbitrary size.

(I). Intermolecular  $N-H\cdots O$  hydrogen bonds form sheets parallel to the *ab* plane (see Table 1 and Fig. 2).

## **Experimental**

See *Comment* and Faul *et al.* (1999) for details. The title compound was obtained by slow evaporation of an ethanol solution.

Crystal data

 $C_{21}H_{18}N_2O_5$   $M_r = 378.38$ Monoclinic, C2/c a = 18.448 (2) Å b = 9.9631 (4) Å c = 20.029 (2) Å  $\beta = 108.422 (5)^{\circ}$   $V = 3492.7 (5) \text{ Å}^3$ Z = 8

#### Data collection

Enraf–Nonius CAD-4 diffractometer  $\omega/2\theta$  scans Absorption correction:  $\psi$  scan (*CORINC*; Dräger & Gattow, 1971)  $T_{min} = 0.647, T_{max} = 0.772$ 3673 measured reflections 3559 independent reflections

#### Refinement

Refinement on  $F^2$   $R[F^2 > 2\sigma(F^2)] = 0.044$   $wR(F^2) = 0.121$  S = 1.103559 reflections 268 parameters H atoms treated by a mixture of independent and constrained refinement  $D_x = 1.439 \text{ Mg m}^{-3}$ Cu K\alpha radiation Cell parameters from 25 reflections  $\theta = 65-73^{\circ}$   $\mu = 0.86 \text{ mm}^{-1}$  T = 193 (2) K Block, yellow  $0.65 \times 0.60 \times 0.30 \text{ mm}$ 

3465 reflections with  $I > 2\sigma(I)$   $R_{int} = 0.061$   $\theta_{max} = 74.3^{\circ}$   $h = 0 \rightarrow 22$   $k = 0 \rightarrow 12$   $l = -25 \rightarrow 23$ 3 standard reflections frequency: 60 min intensity decay: 2%

$$\begin{split} w &= 1/[\sigma^2(F_o^2) + (0.0612P)^2 \\ &+ 3.5777P] \\ \text{where } P &= (F_o^2 + 2F_c^2)/3 \\ (\Delta/\sigma)_{\text{max}} < 0.001 \\ \Delta\rho_{\text{max}} &= 0.30 \text{ e} \text{ Å}^{-3} \\ \Delta\rho_{\text{min}} &= -0.26 \text{ e} \text{ Å}^{-3} \\ \text{Extinction correction: SHELXL97} \\ \text{Extinction coefficient: } 0.00211 (15) \end{split}$$





Part of the packing diagram of (I). Dashed lines indicate hydrogen bonds. Only important H atoms are shown.

#### Table 1

Hydrogen-bonding geometry (Å, °).

$D - H \cdot \cdot \cdot A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - \mathbf{H} \cdots A$
$N6-H6\cdots O23^{i}$ $N10-H10\cdots O21^{ii}$	0.96 0.88	2.03 2.13	2.9005 (17) 2.9320 (17)	150 152
$N10-H10\cdots O23^{ii}$	0.88	2.60	3.2759 (18)	134

Symmetry codes: (i)  $\frac{1}{2} + x, \frac{1}{2} + y, z$ ; (ii) x, 1 + y, z.

All H atoms were located in difference Fourier maps. The range of C-H distances is 0.89–1.01 Å and N-H distances is 0.88–0.96 Å. The  $U_{\rm iso}$  values were refined freely (one per H atom and one per methyl group). The methyl H atoms were allowed to ride but not to rotate.

Data collection: *CAD-4 Software* (Enraf–Nonius, 1989); cell refinement: *CAD-4 Software*; data reduction: *CORINC* (Dräger & Gattow, 1971); program(s) used to solve structure: *SIR92* (Altomare *et al.*, 1994); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *PLATON* (Spek, 2003); software used to prepare material for publication: *SHELXL97*.

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